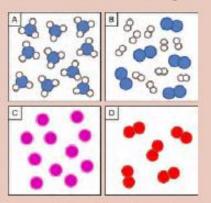
## Grade 9 NS

DATE:\_\_\_\_\_

## <u>Atoms</u>

## Informal Test

Each of the four blocks below (labelled A to E) contain some matter. You
must answer the following questions using the diagrams in the blocks.
Each question may have more than one answer! [12 marks]



- a) Which block represents the particles of an element?
- b) Which block represents the particles in a compound?
- c) Which block represents the particles in a mixture?
- d) Which block represents diatomic particles?
  - e) If the blue atoms are N and the white atoms are H, write the formula for the molecules in block A.
  - f) If the blue atoms are N and the white atoms are H, write the formula for the molecules in block B.
  - g) Which blocks contain molecules?
  - h) Which block contains single atoms?

2. How would you name the following compounds?

- a) Write the name next to each formula in the table below.
- b) Build a model of each compound with play dough.
- c) Draw a picture of one molecule of each compound in the final column of the table.

	m			

Formula of the compound	Name of the compound	Picture of one molecule of the compound
NH3		
CO2		
CuCl <sub>2</sub>		
SO2		

## 3. What are the formulae of the following compounds? [4 marks]

Formula of the compound	Name of the compound
	sodium chloride
	dinitrogen monoxide
	sulfur trioxide
	carbon monoxide